

TETRAHEDRON

Tetrahedron 58 (2002) 8629–8659

Tetrahedron report number 619

Cyclic amino acid derivatives

Kyung-Ho Park^a and Mark J. Kurth^{b,*}

aDuPont, Central Research and Development, Experimental Station, P.O. Box 80328, Wilmington, DE 19880-0328, USA
bDepartment of Chemistry, University of California, Davis, CA 95616, USA ^bDepartment of Chemistry, University of California, Davis, CA 95616, USA

Received 14 June 2002

Contents

1. Introduction

Conformationally constrained amino acids (AAs) have been the focus of both synthetic and medicinal chemistry, particularly as they apply to the design of novel peptides.^{[1](#page-27-0)}

Abbreviations: AAs, amino acids; ACHC, aminocyclohexanecarboxylic acid; ACPC, aminocyclopenatnecarboxylic acid; ACPD, aminocyclopentanedicarboxylic acid; AIBN, 2,2'-azobisisobutyronitrile; BDP, bis(dimethylaluminum)propane-1,3-dithiolate; Bn, benzyl; Boc, tertbutyloxycarbonyl; CAA, cyclic amino acid; Cbz, carbonyloxybenzyl; DCC, 1,3-dicyclohexylcarbodiimide; DDQ, 2,3-dichloro-5,6-dicyano-1,4 benzoquinone; DEAD, diethyl azodicarboxylate; DHCGA, 2,3-dehydrocycloglutamic acid; DIC, diisopropylcarbodiimide; DMAP, 4-dimethylaminopyridine; DME, dimethoxyethane; DMF, dimethylformamide; DMSO, dimethylsulfoxide; DPPA, diphenylphosphorylazide; EAA, excitatory amino acid; EtOAc, ethyl acetate; Fmoc, 9-fluorenylmethoxycarbonyl; GABA, γ-aminobutyric acid; HATU, o -(7-azabenzotriazol-1-yl)-N,N,N',N'-tetramethyluronium hexafluorophosphate; HMPB, 4-(4-hydroxymethyl-3-methoxyphenoxy)-butyric acid; HOAT, 1-hydroxy-7-azabenzotriazole; HOBT, 1-hydroxybenzotrizole; HOSU, N-hydroxysuccinimide; LDA, lithium diisopropylamide; LTMP, lithium 2,2,6,6-tetramethylpiperidine; MBHA resin, 4-methyl benzhydrylamine resin; MCPBA, m-chloroperoxybenzoic acid; Ms, methanesulfonyl; NMO, 4-methylmorphoine N-oxide; PEG, polyethyleneglycol; PLE, pig liver esterase; RCM, ring closing metathesis; TBDMS, tert-butyldimethylsilyl; TBDPS, tert-butyldiphenylsilyl; TEMPO, 2,2,6,6-tetramethyl-1-piperidinyloxy; TFA, trifluoroacetic acid; THF, tetrahydrofuran; TMSCl, trimethylsilylchloride; Ts, p-toluenesulfonyl.

Rigidified cyclic amino acids (CAAs) have also played an important role in drug design and development^{[2](#page-27-0)} where they exert conformational constraints while maintaining the hydrophobic character of the linear alkyl chains.^{[3](#page-27-0)} Thus, the incorporation of CAAs into peptides or peptidomimetics induce conformational restrictions and provide important structural effects. Furthermore, CAAs in peptide surrogates have been used for structural and biomechanistic investigation as well as for probing the structural requirements of receptor-bound ligand conformations. This report provides an overview of the preparation of conformationally constrained CAAs and their derivatives.

2. α -AA derivatives

2.1. Cyclopropane AA derivatives

The cyclopropane ring system is particularly interesting because it affects the chemical and biological properties in peptides through significant conformational restrictions in the AA residues. They are found in a wide class of naturally occurring products, and synthetic 2,3-methanoamino acids are structurally related to those cyclopropane AAs found in proteins. Racemic 2,3-methanoamino acids (4) have been synthesized by several methods as reviewed by Stammer^{[4](#page-27-0)} and Alami.^{[5](#page-27-0)} These methods include (i) dialkylation of glycine or malonate derivatives (1) with ethylene dibromide or its equivalent, (ii) 1,3-dipolar cycloaddition of diazomethane on β -substituted acrylic acid derivatives (3) to

Keywords: cyclic amino acids; α -amino acid; β -amino acid; γ -amino acid; heterocycle; synthesis; preparation.

 $*$ Corresponding author. Fax: $+1-530-752-8192$; e-mail: mjkurth@ucdavis.edu

Scheme 1.

form pyrazolines, and (iii) intramolecular cyclization of γ -substituted α -amino butyric acid derivatives (2) (Scheme 1).

Optically active 2,3-methanoamino acids and their derivatives have also been synthesized using chiral auxiliaries, including Schölkopf's bislactim ethers, $\overline{6}$ $\overline{6}$ $\overline{6}$ Husson's 1,3-oxazolidine, $\frac{7}{1}$ $\frac{7}{1}$ $\frac{7}{1}$ and the diketopiperazine auxiliary.^{[8](#page-27-0)} At the same time, many side-chain functionalized 2,3-methanoamino acids have also been prepared from optically pure lactones or D-mannitol. Some representative 2,3-methanoamino acid derivatives are shown in Figs. 1 and 2; Burgess reviewed these approaches in detail.^{[9](#page-27-0)}

New chiral 2,3-methanoamino acids with extended sidechains were prepared from cyclopropane chiron 6, which was synthesized from diol 5^{10} 5^{10} 5^{10} Thus, diol 5 was reacted with thionyl chloride to give a cyclic sulfite, which was oxidized into a cyclic sulfate and then reacted with diethyl malonate to afford chiron 6. Selective hydrolysis of the less hindered ester in 6 followed by Curtius rearrangement and hydrogenolysis afforded Boc-protected amino alcohol 7. From this intermediate, several Boc-protected 2,3-methanoamino acid derivatives $(8-10)$ were prepared following the sequences shown in [Scheme 2](#page-2-0).

2,3-Methanoamino acids were also synthesized diastereo-

E-cvclo-Met

ation and S'_N cyclization of 1,4-dichlorobut-2-ene by the benzophenone Schiff base of aminoacetonitrile ([Scheme](#page-2-0) [3](#page-2-0)).¹¹ After palladium(0)-catalyzed alkylation of (E,Z) -1,4dichlorobut-2-ene by the Schiff base anion, the reaction proceeded to the π -allyl palladium intermediate 11. The S^{\prime}_N cyclization of the π -allyl palladium intermediate 11 was completely stereoselective, affording the sterically favorable 12 (de=100%) with a syn relationship between the ethyl and nitrile groups. Diimide reduction of the ethenyl substituent followed by acid hydrolysis provided racemic coronamic acid (13).

selectively via the one pot palladium(0)-catalyzed alkyl-

This tandem reaction was also attempted using chiral (4S)- 1-chloropent-2-en-4-ol.^{[12](#page-27-0)} Thus, palladium(0)-catalyzed alkylation of this allylic chloride by the Schiff base followed by S'_N cyclization under Mitsunobu conditions afforded 14 diastereoselectively (88% de). Diimide reduction of 14 followed by hydrolysis provided (1S,2S)-homo-coronamic acid (15) in 84% ee ([Scheme 4\)](#page-2-0).

Chiral imine $(-)$ -16 has also been utilized in the preparation of 2,3-methanoamino acid precursors as outlined in [Scheme](#page-2-0) $5¹³$ $5¹³$ $5¹³$ $5¹³$ The starting imine (-)-16 was prepared by the condensation of $(-)$ - $(1S,2S,5S)$ -2-hydroxy-3-pinanone^{[14](#page-27-0)} with aminoacetonitrile in the presence of BF_3 -etherate in

Boc-Z-cyclo-Arg'(Ts) Fmoc-Z-cyclo-Arg'(Mtr) Boc-Z-cyclo-Arg (Ts)

Figure 2. Side-chain functionalized 2,3-methanoamino acids from lactones or D-mannitol.

Scheme 2. (a) SOCl₂; (b) RuCl₃, NaIO₄; (c) diethyl malonate, NaH; (d) KOH/EtOH; (e) N₃PO(OPh)₂, Et₃N; (f) H₂, Pd/C; (g) MsCl, Et₃N; (h) NaCN/DMF; (i) NaOH/MeOH; (j) Ra/Ni, NH₄OH; (k) CbzCl, Na₂CO₃.

Scheme 3.

benzene. Asymmetric palladium(0)-catalyzed alkylation followed by S_N' cyclization afforded diastereochemically pure $(-)$ -E-17 in 70% yield. Subsequent hydrolysis provided 1-amino-2-vinylcyclopropanecarbonitrile E-18 with 2.4% ee. The same reaction from $(+)$ -16 gave the diastereochemically pure precursor $(+)$ -E-17 in 70% yield and subsequently provided $E-18$ with 13.5% ee. Pd(0)-induced reversible ring opening^{[15](#page-27-0)} of the vinylcyclopropane moiety appears to be responsible for the low enantiomeric excesses obtained.

Kurth recently reported the regio- and diastereoselective synthesis of 2,3-methano amino esters from 4-bromo-benzaldehyde Schiff base without palladium catalyst.^{[16](#page-27-0)} By

Scheme 4.

Scheme 7. (a) LiAlH₄; (b) NaBH₄; (c) TBDMSCl, imidazole, CH₂Cl₂; (d) NalO₄, THF/H₂O; (e) Ph₃P=CHR; (f) HN=NH; (g) RuCl₃-NalO₄; (h) DPPA, BuOH.

 31

 $R = H.40%$

 $R = n-Bu, 65%$

 $R = H 71%$

 $R = n-Bu$, 80%

modifying bis-electrophile 20b, 1-amino-2-vinyl cyclopropane carboxylate ester derivatives 24 were synthesized with the amine and vinyl moieties trans to each other (Scheme 6). The complete *trans* diastereoselectivity results from transition-state selectivity in the ring closing step where the energy difference between the cis and trans geometries for the imine and vinyl substituents is energetically favorable for trans-22. When $cis-1,4$ -dichloro-2-butene 20a (X=Cl) was employed as the alkylating agent, only 1-aminocyclopent-3-ene-1-carboxylate 23 was obtained. On the other hand, $cis-1,4$ -dimesyl-2-butene 20a (X=OMs) delivered both 23 and 24 $(23/24=1:2)$.

32

2,3-Methanoamino acids have also been synthesized by

utilization of the $[2+1]$ cycloaddition reaction of 1-seleno-2-silylethenes (Scheme 7).^{[17](#page-27-0)} This approach to the cyclopropane ring derives from a selenium-stabilized 1,2-silicon migration process.^{[18](#page-27-0)} Thus, the reaction of 1-(phenylseleno)-2-(triethylsilyl)ethane and di-tert-butyl methylenemalonate afforded the selenium- and silicon-containing cyclopane 25 along with cyclobutane byproduct $26 (25/26=7:3, 90-98\%)$ yield). Stereoselective reduction of the less hindered ester functional group transto the (phenylseleno)(triethylsilyl) methyl group and alcohol protection provided the primary alcohol derivative 27. Subsequent oxidation afforded aldehyde 28 and Wittig reaction followed by diimide reduction gave synthetic intermediate 29 (R=H). Sequential deprotection of the TBDMS group, oxidation, and Curtius

30

R

Scheme 8. (a) THF/'BuOK, allyl bromide; (b) hv , acetophenone; (c) 6N HCl.

Scheme 6.

Scheme 9. (d) $(Boc)₂O$, NaOH; (e) $CH₂N₂/Et₂O$; (f) RuO₄; (g) HCl/H₂O; (h) BH₃/THF.

Scheme 10. (i) $SOC1_2/CH_2Cl_2$; (j) Et_3N/CH_2Cl_2 ; (k) K_2CO_3 , MeOH, H_2O .

rearrangement afforded Boc-protected coronamic acid 31 $(R=H)$, which was finally deprotected and hydrolyzed to provide racemic coronamic acid 32 (R=H). This method provides a general approach to (E) -2-alkyl-1-aminocyclopropane-1-carboxylic acid derivatives.

2.2. Cyclobutane AA derivatives

Natural cyclobutane AAs ,^{[19](#page-27-0)} like 2,4-methanoproline 36, 2,4-methanoglutamic acid 39, and their derivatives 41 and 44, have been synthesized from methyl 2-benzamido-3 chloropropionate 33 (Schemes $8-10$).^{[20](#page-27-0)} Starting material 33 was made from serine by known methods^{[21](#page-27-0)} and converted to azahexdiene 34 in one pot by sequential dehydrohalogenation and amide allylation. Irradiation of 34 with a medium pressure Hanovia lamp afforded the desired photoproduct 35, which was hydrolyzed to give natural cyclobutane AA 36 ([Scheme 8](#page-3-0)).

To obtain 2,4-methanoglutamic acid (39), oxidation precursor 37 was prepared from AA 36 by Boc protection followed by diazomethane esterification. Oxidation of 37 with ruthenium tetraoxide using Sharpless conditions^{[22](#page-27-0)} provided acid 38, which was hydrolyzed to deliver natural 2,4-methanoglutamic acid (39). Hydroxy AA 41 was obtained by reduction of carboxylic acid 38 followed by ester hydrolysis (Scheme 9). On the other hand, 2,4-methanopyroglutamic acid (44) was synthesized by ring closure of acid 38 using thionyl chloride followed by ester hydrolysis (Scheme 10).

The synthesis of several Boc-protected cyclobutane amino esters which incorporate isoxazoline heterocycles (e.g. 52) has been reported by Kurth (Scheme 11).^{[23](#page-27-0)} Bis-alkylation of diethylmalonate with 1,3-dihalopropane $45²⁴$ $45²⁴$ $45²⁴$ afforded the cyclobutane skeleton 46, which was partially hydrolyzed to mono acid 47. Subsequent Curtius rearrangement followed

Scheme 13.

by deprotection of the benzyl ether provided Boc-protected hydroxy cyclobutane amino ester 49. Oxidation to ketone 50 followed by Wittig olefination afforded exo-methylene substituted Boc-protected amino ester 51. At this point, diastereoselective nitrile oxide 1,3-dipolar cycloaddition to the methylene moiety provided several isoxazoline-containing Boc-protected cyclobutane amino esters 52. The diastereoselectivity arises from intermolecular hydrogen

 $\left[\right]$

N_C

.CO₂Et

bonding between the Boc-NH and the incoming nitrile oxide dipole.

2.3. Cyclopentane, cyclohexane, and higher membered CAA derivatives

Simple 5- or 6-membered CAA have been synthesized by the Bucherer–Bergs method.^{[25](#page-27-0)} Here a cyclic ketone is

NHAc

CO₂Et

N_C

a. HCI

Figure 3.

Scheme 12.

Scheme 15.

Scheme 16.

Scheme 18.

converted into its spirohydantoin in the presence of potassium cyanide and ammonium carbonate. Base [Ba(OH)₂, 140°C] or acid [60% H₂SO₄, 140°C] hydrolysis of the spirohydantoin then affords the CAA [\(Scheme 12](#page-5-0)).

O'Donnell has established methodology for CAA synthesis by alkylation of the Schiff base of glycine ester in the presence of a phase-transfer catalyst.^{[26](#page-27-0)} To probe the binding pockets of Phe^{[7](#page-27-0)} (S₇) and Phe^{[8](#page-27-0)} (S₈), two important aromatic residues for pharmacological properties of substance P (SP), several 2-indanyl-glycine derivatives were synthesized by Chassaing et al. 27 Using ethyl cyanoacetate as a glycine equivalent, indan-based unnatural CAA derivatives have also been synthesized under phase-transfer catalysis conditions by Kotha et al. 28 28 28 Thus, ethyl cyanoacetate was bis-alkylated with dibromo-o-xylene derivative 53 in the presence of base (K_2CO_3) and PTC catalyst (tetrabutylammonium hydrogen sulfate) to afford isonitrlie derivatives 54. These isonitrile derivatives were subsequently hydrolyzed (HCl/EtOH) to give cyclic amino esters 55 ([Scheme](#page-5-0)

[13](#page-5-0)). Following the same reaction sequence, naphthalene, anthraquinone, quinxaline, and furan derived cyclic amino esters were synthesized from their isonitrile precursors 54 ([Fig. 3](#page-5-0)).

Another approach to indan-based CAA derivatives employs 1,2-exomethylene dienes ([Scheme 14\)](#page-5-0).[29](#page-27-0) Starting dienes 58 and 61 were prepared from iodo derivatives 56 and 59, respectively, and reaction of these dienes with several dienophiles and subsequent aromatization resulted in 5- and 7-membered CAA derivatives (62–67). These reactions were carried out by reacting the diene and dienophile in benzene at ambient temperature, followed by DDO^{[30](#page-27-0)} treatment in benzene by reflux.

Isoxazoline-incorporated cyclopentane amino esters 71 have been synthesized by alkylation of the 4-bromobenzaldehyde Schiff base of glycine with cis-1,4-dichloro-2- butene ([Scheme 15](#page-6-0)).^{[31](#page-27-0)} Subsequent 1,3-dipolar cycloaddition reaction to the alkene in 69 followed by hydrolysis

Scheme 20.

of the imine in 70 afforded isoxazoline-incorporated cyclopentane amino esters 71, which were isolated as regioisomeric ureas 72.

The preparation of unique and conformationally constrained α -AAs from the Grubbs ring closing metathesis (RCM) reaction has been reported (Scheme 16).^{[32](#page-27-0)} Substrates (74) for the RCM reactions were synthesized in a stepwise manner by use of Schöllkopf methodology employing the

chiral auxiliary (2R)-2,5-dihydro-3,6-dimethoxy-2-isopropylpyrazine.[33](#page-27-0) The alkylations were carried out by lithiation and alkylation using allyl bromide, 4-bromo-1-butene, and 5-bromo-1-pentene. While the monoalkylated bislactim ether was obtained in 75–96% diastereomeric excess, the subsequent dialkylated product 74 was obtained in 95% de; a consequence of differential steric effects that the second alkylating agent experiences as it approaches the carbanionic center at the 5-position. Five to seven-membered

Scheme 21. (a) 1.0 mol\% OsO₄, NMO; (b) NaH, Mel, DMF/THF; (c) 0.1 M TFA; (d) 0.1 M TFA; (e) Ac₂O, DMAP, CH₂Cl₂.

rings 75 were synthesized via RCM reaction, but the 8-membered derivative was not obtained. Schöllkopf bislactim ethers were readily hydrolyzed by 0.2 M trifluoroacetic acid to afford unsaturated cyclic amino esters 76.

 α , β -Unsaturated hydroxy or oxo derivatives of α -CAA 80 or 82 were also synthesized from bislactim ether 73 ([Scheme 17\)](#page-6-0). 34 For the synthesis of oxo derivative 80, substrate 73 was monoalkylated, then employed in an aldol condensation to afford a mixture of aldols $(77; n=1 \text{ or } 2)$. The selectivity at the α -aldol carbon was moderate, but the aldol reaction of 73 with acrolein has been reported to result in high stereoselectivity.[35](#page-27-0) Oxo substrate 78 (for 6- to 7-membered ring) was synthesized by Swern oxidation of the allylic alcohol. Subsequent RCM and acid hydrolysis provided 5- to 7- membered cyclic amino esters 80.

The keto moiety seems to have little electronic influence to either substrate and the yields are mostly dependent upon conformational preferences in the two substrates. For the preparation of hydroxy derivatives 82, aldols 77 $(n=1 \text{ or } 2)$ were cyclized via RCM to afford cyclized products 81 which, upon hydrolytic cleavage with TFA, afforded 82. However, efforts to obtain 5-membered rings were not successful. Another attempt to hydrolyze 77 $(n=0)$ prior to the cyclization resulted in the dipeptide 83, which was protected as an N-acetyl derivative to afford 84. RCM reaction of 84 finally provided the 5-membered serine analogue 85 as a dipeptide ([Scheme 18\)](#page-7-0).

Hydroxymethylated α -CAA derivatives were also synthesized employing vinyloxirane as the alkylating agent ([Scheme 19\)](#page-7-0).[36](#page-27-0) By stepwise alkylation of bislactim ether 73 with 4-bromo-1-butene and vinyloxirane (in the presence of BF₃-etherate), hydroxymethyl derivatives 86 and 87 were obtained. Due to steric hindrance, differences between the lithiated homoallyl bislactim ether and the more substituted carbon of the epoxide, attack at the less substituted epoxide carbon predominated, resulting in the β -hydroxy derivative 88.^{[37](#page-27-0)} Cyclization of 86 and 87 via RCM reaction afforded tricyclic spirane 89 and hydroxymethyl heterospirane 90, respectively. Tricyclic spirane 89 was obtained via a re-esterification with expulsion of methanol under the RCM reaction conditions. Due to steric congestion in the bislactim-spiranes, a partial hydrolytic cleavage with ring opening of the bislactim ether at the 3,4-positions or 1,6-lactim positions delivered dipeptides 92 and 93, respectively. However, in the cleavage of tricyclic spirane 89, only the dipeptide 91 was obtained from hydrolysis in the 3,4-lactim position.

Undheim et al. also synthesized α -CAAs via a palladium mediated cyclization.[38](#page-27-0) Thus, bislactim was stepwise alkylated to afford several diene and enyne substrates ([Scheme 20](#page-8-0)). Employing a Heck reaction delivered 3,4-dimethylene derivative 95 from dibromide 94e. Substrates carrying a butenyl alkene substituent (94b and 94d) afforded 3,4-dimethylenecyclohexane 97 almost exclusively via competitive 6-exo and 6-endo Heck reactions. Substrates 94a and 94c delivered 3,4-dimethylenecyclopentane 4-ene and 5-mehtylenecyclohex-3-ene 96 via both 5-exo and 6-endo cyclizations in an intramolecular Heck reaction. The product ratio was dependent upon reaction conditions. On the other hand, the acetate-bridged palladacycle^{[39](#page-27-0)} [available

Scheme 22.

from the reaction of $Pd(OAc)_2$ with tri-*o*-tolylphosphane] converted 94c into 95 and 99 in almost the same ratio $(95/99=1:1.1).$

Conformationally constrained vicinally dihydroxylated CAAs, prepared as rigidified analogs of serine or homoserine, were prepared via the streoselective methodology outlined in [Scheme 21.](#page-9-0)^{[40](#page-27-0)} Spirocycloalkene 75a was oxidized to dihydroxy products 104 and 105 (the ratio, 3:1). Hydrolytic cleavage of the pyrazine ring using 0.1 M TFA in aqueous acetonitrile afforded the highly water soluble methyl amino ester. Consequently, this substrate was isolated as either its dimethoxy analog 106 or its triacetylated AA ester derivative 108.

Stereoselective synthesis of vicinal cis-dihydroxy-1-aminocyclohexane-carboxylic acid methyl esters and their methoxy analogs were also prepared in a manner similar to that described in [Scheme 21](#page-9-0). High stereoselectivity was obtained for the 6-membered ring analogs of 104 and 105 (14:1, respectively).

1-Aminocyclopentane-1,3-dicarboxylate (trans-ACPD) $109⁴¹$ $109⁴¹$ $109⁴¹$ is known to activate the metabotropic receptor, one of several EAA (excitatory AA) receptors. To probe for potential activity against that receptor, several α -CAAs with a second acidic group were synthesized ([Fig. 4\)](#page-9-0). For example, 1-amino-2-cyclohexene-1,3-dicarboxylic acid

(DHCGA) $110₁⁴²$ $110₁⁴²$ $110₁⁴²$ an unsaturated cyclic analog of glutamic acid, was prepared from 3-carboxy-4-cyclohexenone by a Bucherer–Bergs reaction. Conformationally constrained ACPD analogs 111^{43} 111^{43} 111^{43} were synthesized from (\pm) -1-phenyl-bicyclo^[3,3,0]octan-3,7-dione.^{[44](#page-28-0)} On the other hand, cyclopentane- and cyclohexane-derived analogs of glutamic acid 112 were obtained from the corresponding 3-ketocycloalkyl carboxylic acids by a combination of microbial and chemical methods.[45](#page-28-0)

To expand the scope of substituents in 112 , Na j era^{[46](#page-28-0)} synthesized several α -CAAs 113 with a second acidic group (carboxylic acid, phosphonic or tetrazole group) attached to 5-, 6-, or 7-membered ring skeletons (Scheme 22). Thus, cyclic ketones 114[46](#page-28-0) were subjected to Bucherer–Bergs reaction to afford spirohydantoins 115. Upon hydrolysis, these provided AA hydrochlorides 113.

The Diels–Alder reaction provides another useful route to CAA derivatives. Use of 2-phenyl-4-benzyliden- $5(4H)$ oxazolone 116 as dienophile in the Diels–Alder reaction of several dienes affords various cyclic analogs of pharmacologically interesting natural AAs 121 and 122 (Scheme 23).^{[47](#page-28-0)} In particular, 2-aminobicyclo[2,2,1]heptane-2-carboxylic acid has been used to evaluate the transport of AAs with hydrophobic side-chains.^{[48](#page-28-0)} $5(4H)$ -Oxazolones 116 were synthesized by the Erlen-meyer–Plochl method^{[49](#page-28-0)} where the *endolexo* ratio (117/118)

Scheme 24.

was found to be independent of the nature of the substituents.

The spiroxazolone mixture (117 and 118) was separated by an iodolactonization procedure (steps h and i). Methyl esters 119 and 120 were then obtained by esterification with diazomethane followed by hydrogenation using 10% palladium on carbon. The benzamide group in 120 was hydrolyzed with 6N HCl and the free AA 122 was obtained by treatment of the resulting HCl salt with propylene oxide.

 $(+)$ -(2S)-2-Aminobicyclo[2,2,2]-octane-2-carboxylic acid 127, a homologue of 121, was also synthesized^{[50](#page-28-0)} in a study to selectively perturb the levels of neutral AAs in the cerebral cortex. Thus, the Diels–Alder reaction of chiral oxazolidinone 123 and cyclohexadiene afforded diasteromers 124 and 125 which, upon methanolysis, provided ester 126 as a single diastereomer. Diastereomer 125 came from a thermally induced C-2 epimerization of 124 during the Diels–Alder reaction.^{[51](#page-28-0)} Subsequent hydrogenation of 126 followed by acid hydrolysis finally delivered the hydrochloride salt 127 (Scheme 24).

3. **B-AA** derivatives

b-AAs are a widespread class of non-proteinogenic AAs which are found in nature both in free form and in peptides. They are also important elements in some well known enzyme inhibitors, statins, and the anticancer agent taxol. 52 As probes of both structural effects and properties,

conformationally constrained α - and β -AAs have been widely employed in peptide synthesis.^{[53](#page-28-0)}

Maleic anhyrides (128) have been used as starting materials in the enantioselecive synthesis of β -CAAs and afford enantiomeric cyclopropane and cyclobutane β -AA derivatives 132 through a chemoenzymatic approach (Scheme $25)$.⁵⁴ Thus, Fisher esterification of anhydrides afford *meso* diesters 129, which were chemoselectively hydrolyzed by pig liver esterase (PLE) to provide hemiesters 130.^{[55](#page-28-0)} Subsequent synthesis of acyl azides followed by Curtius rearrangement in the presence of benzyl alcohol afforded enantiomeric N-Cbz protected amino esters 132. Hydrolysis and deprotection of $132b$ ($n=2$) delivered 2-aminocyclobutanecarboxylic acids 133b.

Enantiomerically pure ethyl (1R,2R)-2-N-(benzyloxycarbonylamino)-2-methyl-1-cyclopropane carboxylate 138 was synthesized from the unsaturated ester 134 (Scheme 26).^{[54](#page-28-0)} Thus, Wittig–Horner reaction of D-glyceraldehyde with a suitable phosphonate provided the starting ester 134.^{[56](#page-28-0)} A highly stereoselective 1,3-dipolar cycloaddition reaction^{[57](#page-28-0)} of 134 followed by photolysis of the resulting pyrazoline and subsequent hydrolysis of the acetonide delivered cyclopropane derivative 135 in 60% overall yield. Oxidation of the diol afforded acid 136. Its conversion to acyl azide 137, followed by Curtius rearrangement, provided 138 in 53% overall yield from diol 135.

2-Aminocyclobutanecarboxylic acids, which lack the 'GABA structural element', were synthesized following

Scheme 26.

Scheme 27. (i) $Pb(OAc)$ ₄/MeOH/toluene; (ii) (a) NaOH/EtOH, (b) H+.

the reaction sequences in Scheme 27.^{[58](#page-28-0)} Thus, starting *cis*and *trans*-amido esters $(139 \text{ and } 142)^{59}$ $(139 \text{ and } 142)^{59}$ $(139 \text{ and } 142)^{59}$ underwent lead tetraacetate induced rearrangement to carbamates 140 and 143, respectively. These were hydrolyzed to provide *cis*and trans-2-aminocyclobutanecarboxylic acids 141 (56% yield) and 144 (66% yield), respectively.

2-Aminocyclopentanecarboxylic acids (ACPC) and their derivatives have shown a wide range of interesting biological activities. It is known that 2-aminocyclopentanecarboxylic acid containing oligomers adopt a helical

conformation^{[60](#page-28-0)} and their β -peptide displays interesting antibiotic activity. 61 As for the 5-membered cyclic b-aminoacid derivatives, enantiomerically pure trans ACPC esters $(-)$ -148 and $(+)$ -148 were prepared from chiral ketones $(-)$ -145 and $(+)$ -145, respectively (Scheme 28). These AA derivatives were synthesized as replacements for proline in potential HIV protease inhibitors containing a hydroxyethylamine dipeptide isostere. 62 Thus, the ketone moiety in $(-)$ -145 was protected and the diester was monohydrolyzed to afford thioketal 146, which underwent Curtius rearrangement to provide Boc protected amino

Scheme 29. (a) $(Boc)_2O$, NaHCO₃, THF/H₂O; (b) (+)-ephedrine, EtOAc, crystallization; (c) (-)-ephedrine, EtOAc, crystallization; (d) 1 M NaHSO₄; (e) tert-butylamine, HOBt, EDC.

Scheme 30.

ester $(-)$ -147. Reduction of the thioketal using Raney nickel gave Boc protected trans 2-aminocyclopentanecarboxylic ester $(-)$ -148 in 88% yield. Enantiomer $(+)$ -148 was likewise synthesized from $(+)$ -145 following the same sequences shown in [Scheme 28](#page-12-0).

Enantiomerically pure *cis*-ACPC esters $(-)$ -151 and $(+)$ -151 were prepared from racemic *cis*-2-ACPC 149, which was synthesized by a known method 63 [\(Scheme 29\)](#page-12-0). Boc protection of the starting amine afforded racemic cis-150, and subsequent resolution by chiral ephedrine delivered each acid 150 in high enantiomeric excess. The carboxylic acid moiety of each antipode was coupled with tert-butylamine to deliver the two derivatized enantiomers 151.

On the other hand, pure Fmoc protected trans-2-ACPC 156

was prepared from ethyl 2-oxocyclopentanecarboxylate (Scheme 30).⁶⁴ Cyclic β -ketoester 152 was converted into $(1R, 2S)$ -2-hydroxycyclopentanecarboxylate 153 via Baker's yeast reduction. Mitsunobu reaction of 153 with $HN₃$, followed by reduction and Boc protection, afforded trans 2-aminocyclopentanecarboxylic ester 155. Subsequent hydrolysis and Fmoc protection provided (1R,2R)- Fmoc-ACPC 156.

For the other enantiomer, ketoester 152 was reacted with (S) - α -methylbenzylamine to give enamine 157, which was reduced to γ -amino alcohol 158. Removal of the chiral auxiliary followed by Fmoc protection afforded protected γ -amino alcohol 159, which was oxidized to provide (1S,2S)-Fmoc-ACPC 156 (Scheme 31). The development of these synthetic routes to the two enantiomers of Fmoc-ACPC should facilitate the investigations of conformational and biological properties of 12 -helical β -peptides.

Racemic 2-aminocyclohexanecarboxylic acids (ACHC) were prepared either from anthranilic acid by the action of sodium in isoamyl alcohol^{[65](#page-28-0)} or from the reaction of 1-cyclohexene-1-carboxylic acid with ammonium hydroxide.^{[66](#page-28-0)} Pure trans-ACHC^{[67](#page-28-0)} was prepared by resolving the quinine salt of N-benzoyl-trans-ACHC:^{[68](#page-28-0)} the tetramer and the hexamer of this β -AA are known to have helical conformations with 14-membered-ring hydrogen bonds between a carbonyl oxygen and the amide proton of the second residue from the N-terminus. Asymmetric synthesis of a *trans*-ACHC derivative^{[69](#page-28-0)} was accomplished from pyrrolobenzodiazepine-5,11-dione (Scheme 32). Thus,

Scheme 31.

Scheme 33.

alkali metal reduction of 160 in ammonia afforded dione 161, which underwent methanolysis to provide 162. Tosylation of the free amine to give 163, followed by hydrolysis, gave $(+)$ -trans-ACHC derivative 164 in 58% yield.

Enantiomerically pure substituted cis- and trans-2-aminocyclohexanecarboxylic esters 168 and 169 have been synthesized by Barluenga from 4-nitrocyclohexanones 167.^{[70](#page-28-0)} These nitrocyclohexanones were derived from asymmetric cycloadditions of 2-aminodienes 165 and nitroolefins 166 (Scheme 33).

For the synthesis of the *cis*-BAA derivative, the hydroxy group of 170 was deprotected under acidic conditions to afford nitro hydroxy ketone 171 together with its hemiketal 172. Jones oxidation of the mixture followed by diazomethane esterification provided nitroester 173 as a single compound. Subsequent hydrogenation of the nitro group using Raney nickel delivered cis - β -amino ester 174 in 80% overall yield (Scheme 34).

As for the trans derivative, starting material 175, derived from β -(3-furyl)nitroethylene, was employed. Thus, oxidation 71 of the furan ring followed by diazomethane esterification provided β -nitro ester 176. Reduction of the nitro group afforded *trans*- β -amino ester 177 in 91% overall yield (Scheme 35).

4. γ -AA derivatives

2-Aminocycloalkylacetic acids 178, with their restricted rotation about the $C(3)$ and $C(4)$ atoms, have been synthesized^{[72](#page-28-0)} as analogs of the inhibitory neurotransmitter γ -aminobutyric acid (GABA). Thus, cyclopropane (184) and cyclobutane analogs (185) were prepared from readily available diesters $179a^{73}$ $179a^{73}$ $179a^{73}$ and $179b^{74}$ $179b^{74}$ $179b^{74}$ respectively. Sequential partial diester hydrolysis, acid to alcohol reduction, and tosylation of the resulting alcohols delivered tosylates 180a and 180b. Upon treatment with potassium cyanide, nitriles 181a and 181b were obtained. These were converted into acid hydrazides $(182a$ and $182b)^{75}$ $182b)^{75}$ $182b)^{75}$ which in

Scheme 34.

Scheme 36. (i) NaOH/EtOH, then BH₃·THF/Et₂O, then TSCl/pyridine; (ii) KCN/DMSO; (iii) hydrazine/EtOH; (iv) NaNO₂, then heating, then MeOH; (v) NaOH/EtOH.

Scheme 37.

turn delivered carbamates (183a and 183b) via Curtius rearrangement. Carbamate hydrolysis finally provided the trans-AAs 184 and 185 (Scheme 36). The cis-cyclopropane derivative 187^{[73](#page-28-0)} was synthesized from *cis*-diester 186 following the same sequences as in Scheme 36 (Scheme 37).

On the other hand, cis cyclobutane derivative 191 was prepared from the bromomethyl derivative 188.^{[73](#page-28-0)} However, treatment of compound 188 with potassium cyanide resulted in cleavage of the cyclobutane ring to generate an iminium ion. To prevent this, the amine functional group was protected as the carbamate (189) and subsequent introduction of the nitrile group proceeded smoothly to provide 190. Upon hydrolysis of 190, the desired cis - γ -aminobutyric acid 191 was obtained (Scheme 38).

cis- and trans-Cyclopentyl and cyclohexyl AAs were synthesized from oximes ([Scheme 39](#page-16-0)). For cyclopentyl derivatives, oxime 192 was hydrogenated over platinum oxide in ethanolic HCl to afford cis-lactam 193 and the trans-amino ester 194, which were subsequently hydrolyzed to provide the cis- and trans-AAs 195 and 196, respectively. The same method was applied to the synthesis of the cyclohexyl AAs. Thus, both cis- and trans-lactams were obtained from hydrogenation of the starting oxime 197. Hydrolysis of these lactams yielded AAs 200 and 201, respectively.

2-(Aminomethyl)cycloalkanecarboxylic acids 205 were also synthesized to restrict bond rotation about the $C(2)$ – $C(3)$ bond of GABA ([Scheme 40](#page-16-0)).^{[76](#page-28-0)} In this work, the synthesis of trans-cyclopropyl and cyclobutyl aminoacid 205a and 205b starts from the Gabriel reaction of tosylates 202 with potassium phthalimide. Dephthaloylation followed by hydrolysis afforded the desired trans products 205, which proved to be potent in GABA receptor binding studies. The synthesis of cis-cycloalkylamino acids were accomplished by reaction of lactones 206^{73} 206^{73} 206^{73} with potassium phthalimide followed by dephthaloylation [\(Scheme 41](#page-16-0)).

Optically active N-protected 1,3-cyclobutane AAs 213 and ent-213 were synthesized from α -pinene [\(Scheme 42\)](#page-17-0);^{[77](#page-28-0)}

Scheme 38. (i) ClCO₂Me/pyridine/Et₂O; (ii) KCN/DMSO; (iii) NaOH/EtOH.

Scheme 39.

Scheme 40. (i) Potassium phthalimide/DMF; (ii) MeNH₂/EtOH; (iii) NaOH/EtOH.

these AAs have potential applications in designed peptidomimetics. The starting material, verbenone, obtained from allylic oxidation of $(+)$ - α -pinene, was oxidatively cleaved to afford keto-acid 209. Subsequent benzylation followed by the haloform reaction delivered acid 211, which underwent Curtius rearrangement to provide Boc-protected amine 212. Debenzylation by hydrogenolysis finally afforded N-protected AA 213 which is suitable for peptide synthesis. To obtain its enantiomer, ent-213, Curtius rearrangement was performed first to give compound 214 and subsequent haloform reaction provided ent-213.

The more flexible cyclobutane AA derivative 216 was synthesized directly from $(+)$ - α -pinene and without oxi-dation to verbenone [\(Scheme 43\)](#page-17-0). Thus, oxidation of $(+)$ - α - pinene gave ketoacid 215, which underwent sequential Curtius and haloform reactions to provide N-protected AA derivative 216. These AAs (ent-213 and 216) were converted into their FMOC protected AAs for solid-phase peptide/peptidomimetic syntheses.[78](#page-28-0)

5. CAA derivatives with the amine group inside the cyclic system

Aziridine-2-carboxylic acids are important components for the preparation of a variety of conformationally constrained peptidomimetic building blocks. These chiral aziridines are also key intermediates in the asymmetric synthesis of α -methyl cysteine, α , β -dimethylcysteines and

Scheme 41. (i) Potassium phthalimide/DMF; (ii) MeNH₂/EtOH.

Scheme 42.

CO₂Bn

218

 $NaN₃$

НŃ

99%

CO₂Bn

220

OH

CO₂Bn

219

 $\left[\frac{PPh_3}{CH_3CN}\right]$ reflux

 $83%$

Scheme 43.

Scheme 44.

∩

217

OН

a. NaIO₄ RuCl₃

b. DCC, DMAP.

B_{nOH}

63%

Scheme 46.

 α, β -dimethyltryptophans.^{[79](#page-28-0)} The (R)-2-methylglycidol 217, which was obtained from the corresponding allylic alcohol via Sharpless asymmetric epoxidation, was oxidized to the acid. Subsequent formation of benzyl ester 218 followed by ring opening by azide provided azido alcohol 219. Enantiomerically pure aziridine ester derivative 220 was obtained by refluxing 219 with triphenylphospine ([Scheme 44\)](#page-17-0).

Cbz protected aziridne 224, an important synthon of α, β -dimethylcysteines or α, β -dimethyltryptophans, was synthesized from benzyl tiglate [\(Scheme 45\)](#page-17-0). Thus, diol 221 was prepared by Sharpless asymmetric dihydroxylation with AD-mix- α . 2,3-Cyclic sulfite formation followed by oxidation afforded cyclic sulfate 222, which provided α -azido ester 223 by azide substitution followed by acidic hydrolysis. α -Azido ester 223 was then stereospecifically transformed to Cbz protected aziridne 224.

Another substituted aziridine-2-carboxylate (230) was synthesized by reaction of $(S)-(+)$ -benzylidene-p-toluenesulfinamide 225 with the lithium enolate of methyl α -bromopropionate (Scheme 46). Thus, trans-(2R,3S)-(+)-227 was obtained as the major product by Darzens-type condensation.[80](#page-28-0) Oxidation to compound 227 followed by sequential treatment with LDA and MeMgBr afforded substituted aziridine-2-carboxylate 230.[81](#page-28-0)

Azetidine-2-carboxylic acid was first isolated from Convallaria majalis (lily-of-the-valley) in 1955 , 82 and also synthesized by Fowden in 1956.^{[83](#page-28-0)} It is an important constituent in several natural products^{[84](#page-28-0)} and is used for the investigation of secondary structure in unnatural poly-peptides.^{[85](#page-28-0)} Fowden synthesized it from γ -aminobutyric acid. Thus, Hell–Volhard–Zelinsky reaction followed by cyclization provided racemic azetidine-2-carboxylic acid 231 in very low yield (Scheme 47). It was also synthesized as a racemate by Cromwell 86 86 86 from 2,4-dibromobutyric acid.

An enantioselective synthesis of L-azetidine-2-carboxylic acid and its derivatives was reported by Hanessian et al. (Scheme 48).^{[87](#page-28-0)} In this work, the Oppolzer sultam derivative of glyoxylic acid oxime 232 was reacted with allyl bromide to give chiral allyl glycine 233. Subsequent hydrolysis of the

Scheme 47.

Scheme 48. (a) Allyl bromide, Zn, NH₄Cl; (b) LiOH/THF; (c) CH₂N₂-Et₂O, MeOH; (d) Cbz-Cl, NaHCO₃; (e) O₃, MeOH; (f) NaBH₄; (g) MsCl, pyridine; (h) H_2 , 10% Pd/C; (i) NaHCO₃; (j) 3N HCl, Dowex.

Scheme 49.

Scheme 50.

chiral auxiliary followed by esterification and Cbz protection afforded the methyl ester 234. Ozonolysis followed by aldehyde reduction and mesylation delivered mesylate 235, which was deprotected and cyclized to give L-azetidine-2 carboxylic acid methyl ester 236. Ester hydrolysis and Dowex purification delivered L-azetidine-2-carboxylic acid 237 in 15% overall yield. Isopropyl, cinnamyl and naphthyl substituted azetidine-2-carboxylic acids were also prepared enantiomerically pure by the same reaction sequences.

Seebach reported a method for the alkylation of 2-azetidinecarboxylic acid at the 2 position without racemization (Scheme 49).[88](#page-28-0) Thus, labile protons in 2-azetidinecarboxylic acid were replaced with trimethylsilyl groups to afford 238, which in turn reacted with pivalaldehyde to give ester 239. Treatment of 239 with LDA followed by aldehyde addition provided 240 diastereoselectively $(>96\%$ ds).

The synthesis of isomeric azetidine-3-carboxylic acid 244 was also reported (Scheme 50).^{[89](#page-28-0)} The starting material, 1-benzhydrylazetidin-3-ol (241) , 90 was mesylated and then converted into nitrile 242. Subsequent hydrolysis to give 243, followed by debenzylation, afforded azetidine-3 carboxylic acid 244 as a racemic mixture.

There are numerous approaches to the syntheses of substituted proline derivatives. Proline is known to be the major contributor to the biological activity of several proteins and plays an important role in biological recognition phenomena because of its ability to induce β -turns and initiate the folding of an α -helix.^{[91](#page-28-0)} Recently, (2S,3S,4R)-3,4-dihydroxyproline 252, a known potential glycosidase inhibitor, 92 was synthesized from the azido ketene-S,S-acetal 247.^{[93](#page-28-0)} Thus, 2,3-O-isopropylidene-D-ery-throse 245 was converted into 246 by Peterson olefination^{[94](#page-28-0)}

Scheme 51. (a) 2-Lithio-2-trimethylsilyl-1,3-dithiane, -78°C to rt; (b) (PhO)₂PO·N₃, DEAD, Ph₃P; (c) *n*-octane, reflux; (d) NaBH₄, MeOH; (e) (Boc)₂O, Et₃N; (f) TI(OCOCF₃)₃; (g) NalO₄, RuCl₃, H₂O; (h) TFA, DOWEX 50x 8-100.

Scheme 52.

and the requisite azide functional group was introduced from alcohol in 246 using diphenyl phosphoryl azide. Intramolecular 1,3-dipolar cycloaddition between the ketene-S,S-acetal and azide moieties was effected by thermolysis to afford 2-amino ketene-S,S-acetal 248—a masked substituted proline—which was reduced and then protected to afford Boc derivative 249. Hydrolysis of the dithioacetal followed by oxidation of the resulting aldehyde 250 delivered the protected AA 251 without epimerization at C-2. Acid-catalyzed cleavage of both acetonide and Boc groups finally provided proline derivative 252 ([Scheme 51\)](#page-19-0).

Racemic proline derivative 258 was also synthesized by an intramolecular 1,3-dipolar cycloaddition reaction between ketene-S,S-acetal and azide.^{[95](#page-28-0)} The synthesis of ω -azido ketene-S,S-acetals 254 starts from bromoester 253 from which the azide functional group can be introduced (Scheme 52). Bis(dimethylaluminum)propane-1,3-dithiolate $(BDP)^{96}$ $(BDP)^{96}$ $(BDP)^{96}$ was used to protect ester group in the presence of azide functional moiety and subsequent 1,3-dipolar cycloaddition provided the cyclic imine 255. N-Acylation gave Cbzprotected 2-amino ketene-S,S-acetal 256 in 65% overall yield from 254. Treatment with 50% aqueous acetic acid containing a small amount of concentrated hydrochloric

acid afforded acyl sulfide 257 and subsequent sodium methoxide treatment provided racemic proline derivative 258 in 84% overall yield from 256.

Pipecolic acid derivative 262 was also prepared (Scheme 53) from azido ketene-S, S-acetals 254 $(n=2, R=H)$. Thermolysis afforded the cyclic imine which was in turn reduced by $NabH_4$ to give 259 (58% yield from 254). The N-toluenesulfonyl derivative (260) was prepared and its thioacetal functional group was cleaved using PhI(OCOCF₃)^{2[97](#page-28-0)} to give aldehyde **261**. Upon oxidation of 261, pipecolic acid derivative 262 was obtained in 53% overall yield from 260.

Substituted proline derivatives have also been synthesized by transition metal catalyzed chloride transfer cyclizations of carbon-centered glycine radicals (Scheme 54).^{[98](#page-28-0)} The substrates, α -chloroglycine derivatives 265, are synthesized by the reaction of N-monosubstituted carbamates 263 with methyl glyoxylate. Stable hemiacetals 264 react with PCl₅ to provide the requisite radical precursors 265 (84–100%) which, in the presence of Cu(bpy)Cl, deliver the relatively stable captodative radicals 267 en route to substituted prolines 268.

Scheme 55.

While monosubstituted alkenes (for example 269) provided both *exo*-and *endo-cyclized* products (270 and 271), disubstituted alkenes (for example 272) delivered only 5-exo cyclized products (273 and 274; Scheme 55). Ring size and the E/Z-nature of double bond have little impact on the stereochemistry of the cyclization. 2,3-trans-Substituted pyrrolidines are obtained as the major product because of the preferred quasi-axial orientation of the ester substituent in the reaction's chair-like transition state.^{[99](#page-28-0)}

Scheme 56.

Similar results were obtained in the $Bu₃SnH-mediated$ cyclization of α -(phenylthio)glycine derivative 275 to 3-substituted proline derivative 276 (Scheme 56).[100](#page-28-0) These reductive cyclizations also provided mainly the 2,3-transsubstituted proline derivatives, and the phenylthio analogs of 269 also afforded 6-endo cyclization products (277–279). Electronically unbiased 1,2-disubstituted alkenes afford exclusive 5-*exo* cyclization products (280, 281; Fig. 5).

Use of tin tetrachloride in the cyclization of N-(3-alkenyl)- N-(methoxycarbonyl)acetoxyglycine ester 282 delivered trans 4-chloropipecolic ester 284 as the major product via a cationic π -cyclization reaction (Scheme 57).^{[101](#page-28-0)} This transformation is mediated by a rapid cationic aza-Cope of the incipient iminium cation followed by formation of bicyclic dioxycarbenium 283.

Ellman et al. synthesized C-3 substituted pipecolic acid 290 from the N-allylated AA 289 via sequential ring closing

Scheme 58. (a) Acetonyltriphenylphosphonium chloride, Na₂CO₃, dioxane/H₂O reflux, 24 h; (b) catecholborane, CBS-(S)-oxazaborolidene, toluene, -78° C; (c) N-Boc-Gly, DMAP, DIC, CH₂Cl₂; (d) LDA, THF (-20 to -70°C), then ZnCl₂/THF; (e) allyl iodide, NaH, THF; (f) Grubbs catalyst, CH₂Cl₂; (g) 10% Pt/C, H2, EtOAc, 50 psi.

metathesis and hydrogenation (Scheme 58).¹⁰² Thus, Wittig reaction of aldehyde 285 gave E - α , β -unsaturated ketone 286 and subsequent asymmetric reduction of the ketone followed by coupling with N-Boc Gly afforded chiral ester 287. Claisen rearrangement of this ester set the two contiguous stereocenters in AA 288 and selective electrophilic allylation of the Boc-nitrogen in the presence of the carboxylic acid with allyl iodide and NaH in THF gave 289.

Lamaty et al. synthesized various CAA derivatives including pipecolic acid derivatives—via RCM on a polyethylene glylcol (PEG) support (Scheme 59).¹⁰³ An N-tosyl group was used to protect/activate the amino group for N-allylation, but the resulting sulfonamide was sensitive to racemization. Consequently, the mild Mitsunobu reaction was selected to couple 291 with PEG to give 292. Alkylation followed by RCM (using Grubbs' catalyst) afforded PEG-supported CAA derivative 294. Acid

hydrolysis (6N HCl, reflux 4 h) of PEG-ester 294 delivered AA 295 with no racemization. Other PEG supported CAA derivatives 296 (e.g. R=methyl or vinyl when $n=1$, and R=H when $n=2$ or 3) were also prepared by appropriate selection of the alkylating agent.

Chiral pipecolinic acids were also synthesized from α,ω diamino carboxylic acids by a one-step photocatalytic cyclization^{[104](#page-29-0)} (Scheme 60). Thus, L-lysine or its derivatives 297, when photolized $(>300 \text{ nm}, \text{high-pressure Hg arc})$ in the presence of catalytic $TiO₂/PtO₂$, provided pipecolinic acid 298 in $7-43\%$ yields with $13-100\%$ ee. In this reaction, L-lysine undergoes oxidation to give either an α -keto or ω -aldehyde intermediate. Subsequent intramolecular condensation followed by imine reduction provides either an optically pure or a racemic pipecolinic acid.

Corey et al. prepared diastereomeric 3-hydroxy-(S)-pipecolic

Scheme 60.

Scheme 61.

acids 305 and 306 from β -hydroxy- α -amino esters by an aldol coupling reaction (Scheme 61).^{[105](#page-29-0)} Thus, aldol coupling of Shiff base 299 with 4-chlorobutyraldehyde in the presence of the chiral quaternary ammonium salt 300 as catalyst produced a 1:1 ratio of syn and anti adol products 301 and 302. Subsequent cyclization afforded amino esters 303 and 304; hydrolysis provided the corresponding pipecolic acids. The highest syn/anti ratio (13:1) was obtained in the aldol condensation of cyclohexanecarboxaldehyde.

The synthesis of phenylalanine derivatized peptides as

novel N-type calcium channel blockers have been prepared from N-alkylated pipecolinic acids, in turn prepared from substituted pyridine-2-carboxylic acids.^{[106](#page-29-0)} Thus, reduction of pyridine ring in picolinic acid followed by reductive N-alkylation using, e.g. isovaleraldehyde affords 307 (Scheme 62). Pyridine-3- or -4-substituted carboxylic acid as well as 2-pyrazine carboxylic acid, also afford the corresponding pipecolinic acid or 2-piperidine carboxylic acid derivatives, respectively.

Fluorine-containing AAs have received considerable attention as candidates for peptide modification since

Scheme 63.

lipophilicity and electronegativity in the fluorine-containing moiety may improve drawbacks of peptide drugs; e.g. rapid degradation by proteases and low lipophilicity.^{[107](#page-29-0)} Difluoromethylated AAs are important members of the fluorinated family of AAs. Uneyama et al. 108 has reported the synthesis of 309, a new type of fluorinated CAA, and related esters by an intramolecular defluorinative cyclization under basic conditions (LTMP or n -BuLi; Scheme 63). A proposed mechanism for this reaction is depicted in Scheme 64. Thus, intramolecular nucleophilic attack of the carbanion in 310 followed by nucleophilic ring opening of the aziridine in 311 with concomitant defluorination delivers quinazoline 312. Base-mediated proton migration finally affords quinazolic products 309. These are new heteroatom

substituted AA derivatives which may be candidates for biologically important inhibitors.

 α -Trifluoromethyl substituted dehydroproline 316 and dehydropipecolinic acid 317 were synthesized from α -trifluoromethyl- α -amino esters containing two terminal alkenes 315 by ring closing metathesis (Scheme 65).^{[109](#page-29-0)} Thus, imine 313 reacted with vinylmagnesium or allymagnesium bromide to afford 314 and subsequent N-alkylation with allyl bromide gave RCM precursors 315.

The synthesis of 7-membered α -aminocycloalkane carboxylic acids has been achieved from enantiomerically pure glycine derivatives (R) - and (S) -tert-butyl 2-(tert-butyl)-3-methyl-4-oxo-1-imidazolidinecarboxylate (Boc-BMI).^{[110](#page-29-0)} Commercially available (S)-Boc-BMI 318 was stepwise dialkylated to give geminally 5,5-disubstituted imidazolidinones 320. TFA deprotection of the Boc group followed by intramolecular cyclization afforded bicyclic derivatives 322. Subsequent hydrolysis of these cyclized compounds followed by ion-exchange chromatography provided the cyclic amino acids 323 in 70–90% yields [\(Scheme 66](#page-25-0)).

Highly functionalized 7-membered cyclic α -amino acid 330 has also been prepared in the course of the synthesis of (2S,4S,6S)-2-amino-4,6-dihydroxypimelic acid 329 (Scheme $67)$.^{[111](#page-29-0)} Thus, a SnCl₄-catalyzed carbonyl-ene

Scheme 64.

Scheme 65. (a) CH_2 =CHCH₂MgBr/THF or CH₂=CHMgBr/THF, then HCl; (b) NaH, DMF, allyl bromide.

Scheme 66.

Scheme 67.

reaction of 5-(2-propenyl)-imidazolidinone 324 with butyl glyoxylate resulted in a mixture of ene product 326, lactone 327, and rearranged ene product 328. Via chair-like transition state 325, the C–C bond is formed by a $re-si$ interaction between the allylic double bond and glyoxylate carbonyl group. Acidic hydrolysis of both 327 and 328, followed by ion exchange chromatography, provided highly functionalized amino acid 329 (74–86%) as well as minor amounts (15%) of 7-membered cyclic α -amino acid 330.

Favorskii rearrangement^{[112](#page-29-0)} of α -halocycloalkanones is another method for the preparation of cyclic α -amino acid derivatives (Scheme 68).^{[113](#page-29-0)} Thus, starting cyclic ketones 331 were converted into alicyclic ketoximes 332, which in turn underwent a Beckmann rearrangement in the presence of polyphosphoric acid (PPA) to afford the corresponding lactams 333. These lactams were dihalogenated to give α , α -dihalolactams 334, which were then converted into mono halogenated lactams 335 via catalytic hydrogenolysis.

Scheme 69. (a) 2,4-Dimethoxybenzaldehyde, NaBH(Oac)₃, >95%; (b) Boc-allyglycine-OH, HATU, HOAT, NEM, 75%; (c) Grubbs catalyst, reflux, DCM, 60 h, 80%; (d) (1) LiOH, MeOH/H2O, quant; (2) 10% TFA, DCM, 3 h; (3) 1N HCl and lyophilize, 78%; (e) Pd/C, H2, quant.

The homologous series of cyclic α -amino acids 336 were finally obtained through the smooth Favorskii rearrangement of these halolactams.

Eight-membered cyclic pseudo-dipeptide 340 has been synthesized using ring closing metathesis (Scheme 69).^{[114](#page-29-0)} This 8-membered ring dipeptide may be used as an important scaffold^{[115](#page-29-0)} in peptidomimetic research and for studies involving amide self association.^{[116](#page-29-0)} The starting material, (S)-allylglycine methyl ester 337, was converted into compound 338 by reductive amination with 2,4-dimethoxybenzaldehyde. Amide bond formation with Boc- (S)-allylglycine provided compound 339, which underwent a RCM reaction to afford protected olefin 340 in 80% yield. Ester hydrolysis (LiOH), followed by deprotection of the Boc and dimethoxybenzyl group, gave pseudo-dipeptide 341. Reduction of this dipeptide provided saturated pseudodipeptide 342 in 60% overall yield from starting material 337.

Peptide cyclization provides conformationally constrained analogs for probing ligand–receptor interactions and

Table 1. Cyclic peptides synthesized by tyrosine side-chain attachment and resin-bound peptide cyclization

348 $Cyclo-(\beta-Ala-Tyr-Pro-Ser-Lys-\beta-Ala-Arg-Gln-Arg-Tyr)$ Cyclo-(Ahx-Tyr-Pro-Ser-Lys-Ahx-Arg-Gln-Arg-Tyr) 349 Cyclo-(Ala-Aib-Tyr-Pro-Ser-Lys-Ala-Aib-Arg-Gln-Arg-Tyr) 350 ^a The tyrosine side-chain that was attached to the resin is given in bold. PPh ₃	Peptide	Number of backbone atoms
		32 38 36
DEAD ЮH Fmoc-Tyr-OCH ₃		

structure–activity relationships. 117 Recently, Beck-Sickinger et al. developed a solid-phase synthesis of cyclopeptides by AA side-chain attachment to the resin.^{[118](#page-29-0)} Three cyclic analogs of neuropeptides 119 have been synthesized to investigate conformational requirements for receptor affinity and selectivity [\(Table 1](#page-26-0)). Thus, HMPB– MBHA resin or Wang resin 343 was coupled with N-Fmoctyrosine methyl ester by Mitsunobu esterification. Chain assembly of the desired linear peptide methyl ester followed by saponification and subsequent head-to-tail cyclization afforded resin bound cyclic peptide 347. TFA-mediated resin release gave the targeted cyclic peptides 348–350 ([Scheme 70](#page-26-0)).

6. Conclusion

Rigidified α -AAs have played important roles in drug design as well as in the development to effect conformational constraints in peptides. CAA derivatives have been especially interesting targets in peptide-based therapeutics and medicinal chemistry. When incorporated into a peptide chain, these AAs mediate significant changes in peptide conformation, thus facilitating the probe of structural requirements of receptor-bound ligand conformations by affecting the ability of these peptides to fit an enzyme active site. As presented in this review, numerous synthetic methods have been developed to provide a wide range of diverse CAA derivatives.

References

- 1. (a) Cativiela, C.; Díaz-de-Villegas, M. D. Tetrahedron: Asymmetry 1998, 9, 3517–3599. (b) Cativiela, C.; Díaz-de-Villegas, M. D. Tetrahedron: Asymmetry 2000, 11, 645–732.
- 2. (a) Degrado, W. F. Adv. Protein Chem. 1988, 39, 51–124. (b) Goodman, M.; Shao, H. Pure Appl. Chem. 1996, 68, 1303–1308.
- 3. Rovero, P.; Pellegrini, M.; Fenza, A. D.; Meini, S.; Quartara, L.; Maggi, C. A.; Formaggio, F.; Toniolo, C.; Mierke, D. F. J. Med. Chem. 2001, 44, 274–278.
- 4. Stammer, C. H. Tetrahedron 1990, 46, 2231–2254.
- 5. Alimi, A.; Calmes, M.; Daunis, J.; Jacquier, R. Bull. Soc. Chim. Fr. 1993, 130, 5–24.
- 6. Groth, U.; Halfbrodt, W.; Scöllkopf, U. Liebigs Ann. Chem. 1992, 351–355.
- 7. (a) Aitken, D. J.; Royer, J.; Husson, H. Tetrahedron Lett. 1988, 29, 3315–3318. (b) Aitken, D. J.; Royer, J.; Husson, H. J. Org. Chem. 1990, 55, 2814–2820.
- 8. Alcaraz, C.; Herrero, A.; Macro, J. L.; Fernandez-Alverez, E.; Bernabe, M. Tetrahedron Lett. 1992, 33, 5605–5608.
- 9. Burgess, K.; Ho, K.-K.; Moye-Sherman, D. Synlett 1994, 575–583.
- 10. Burgess, K.; Lim, D. J. Org. Chem. 1997, 62, 9382–9384.
- 11. Gaucher, A.; Dorizon, P.; Ollivier, J.; Salaün, J. Tetrahedron Lett. 1995, 36, 2979–2982.
- 12. Dorizon, P.; Ollivier, J.; Salaün, J. Synlett 1996, 1071-1075.
- 13. Dorizon, P.; Su, G.; Ludvig, G.; Nikitina, L.; Ollivier, J.; Salaün, J. Synlett 1998, 483-486.
- 14. (a) Carson, R. G.; Pierce, J. K. J. Org. Chem. 1971, 36,

2319–2324. (b) Yamada, S. I.; Oguri, T.; Shioiri, T. J. Chem. Soc., Chem. Commun. 1976, 136–137.

- 15. Burgess, K. Tetrahedron Lett. 1985, 26, 3049–3052.
- 16. Park, K.-H.; Kurth, T. M.; Kurth, M. J. Tetrahedron Lett. 2001, 42, 991–992.
- 17. Yamazaki, S.; Inoue, T.; Hamada, T.; Takada, T. J. Org. Chem. 2001, 64, 282–286.
- 18. Yamazaki, S.; Tanaka, M.; Yamaguchi, A.; Yamabe, S. J. Am. Chem. Soc. 1994, 116, 2356–2365.
- 19. Bell, E. A.; Querishi, M. Y.; Pryce, R. J.; Jansen, D. H.; Lemke, P.; Clardy, J. J. Am. Chem. Soc. 1980, 102, 1409–1412.
- 20. Hughes, P.; Clardy, J. J. Org. Chem. 1988, 53, 4793-4796.
- 21. Painter, E. P. J. Am. Chem. Soc. 1947, 69, 229-232.
- 22. Carlsen, P. H. J.; Katsuki, T.; Martin, V. S.; Sharpless, K. B. J. Org. Chem. 1981, 46, 3936–3938.
- 23. Park, K.-H.; Kurth, M. J. J. Org. Chem. 2000, 65, 3520–3524.
- 24. Michejda, C. D.; Comnick, R. W. J. Org. Chem. 1975, 40, 1046–1050.
- 25. (a) Avindano, C.; Trigo, G. G. The Chemistry of Hydantoins. Advances in Heterocyclic Chemistry, Academic: New York, 1985; Vol. 38. p 177. (b) Kubik, S.; Meissner, R. S.; Rebeck, Jr. J. Tetrahedron Lett. 1994, 35, 6635–6638.
- 26. O'Donnell, M. J.; Boniece, J. M.; Earp, S. E. Tetrahedron Lett. 1978, 19, 2641–2644.
- 27. Josien, H.; Lavielle, S.; Brunissen, A.; Saffroy, M.; Torrens, Y.; Beaujouan, J.-C.; Glowinski, J.; Chassaing, G. J. Med. Chem. 1994, 37, 1586–1601.
- 28. Kotha, S.; Brahmachary, E. J. Org. Chem. 2000, 65, 1359–1365.
- 29. Kotha, S.; Brahmachary, E.; Sreenivasachary, N. Tetrahedron Lett. 1998, 39, 4095–4098.
- 30. Fu, P. P.; Harvey, R. G. Chem. Rev. 1978, 78, 317–361.
- 31. Park, K.-H.; Olmstead, M. M.; Kurth, M. J. J. Org. Chem. 1998, 63, 113–117.
- 32. Hammer, K.; Undheim, K. Tetrahedron 1997, 53, 2309–2322.
- 33. (a) Schöllkopf, U. Tetrahedron 1983, 39, 2085-2091. (b) Schöllkopf, U.; Hinrichs, R.; Lonsky, R. Angew. Chem. Int. Ed. Engl. 1987, 26, 143–145.
- 34. (a) Hammer, K.; Undheim, K. Tetrahedron 1997, 53, 5925–5936. (b) Krikstolaityte, S.; Hammer, K.; Undheim, K. Tetrahedron Lett. 1998, 39, 7595–7598.
- 35. Schöllkopf, U.; Bardenhagen, J. Liebigs Ann. Chem. 1987, 393–397.
- 36. Hammer, K.; Undheim, K. Tetrahedron: Asymmetry 1998, 9, 2359–2368.
- 37. Gull, R.; Schöllkopf, U. Synthesis 1985, 1052-1055.
- 38. Møller, B.; Undheim, K. Tetrahedron 1998, 54, 5789–5804.
- 39. (a) Herrmann, W. A.; Brossmer, C.; Ofele, K.; Reisinger, C.-P.; Priermeier, T.; Beller, M.; Fischer, H. Angew. Chem. Int. Ed. Engl. 1995, 34, 1844–1848. (b) Beller, M.; Riermeier, T. Tetrahedron Lett. 1996, 37, 6535–6538.
- 40. Hammer, K.; Wang, J.; Flank-Pedersen, M. L.; Rømming, C.; Undheim, K. J. Chem. Soc., Perkin Trans. 1 2000, 1691–1695.
- 41. Schoepp, D.; Bockaert, J.; Sladeczek, F. TiPS 1990, 11, 508–515.
- 42. Trigalo, F.; Acher, F.; Azerad, R. Tetrahedron 1990, 46, 5203–5212.
- 43. Ezquerra, J.; Yruretagoyena, B. Tetrahedron 1995, 51, 3271–3278.
- 44. Bertz, S. H.; Cook, J. M.; Gawish, A.; Weiss, U. Org. Synth. 1986, 64, 27–370.
- 45. Trigalo, F.; Buisson, D.; Azerad, R. Tetrahedron Lett. 1988, 29, 6109–6112.
- 46. Alonso, F.; Micó, I.; Nájera, C.; Sansano, J. M.; Yus, M. Tetrahedron 1995, 51, 10259–10280.
- 47. Avenoza, A.; Busto, J. H.; Paris, M.; Peregrina, J. M.; Cativiela, C. J. Heterocycl. Chem. 1997, 34, 1099–1110.
- 48. Christensen, H. N.; Handlogten, M. E.; Lam, I.; Tager, H. S.; Zand, R. J. Biol. Chem. 1969, 224, 1510–1520.
- 49. Carter, H. E. Org. React. 1946, 3, 198–239.
- 50. Pyne, S. G.; Safaei, G. J. Chem. Res. (S) 1996, 160–161.
- 51. Pyne, S. G.; Safaei, G. J.; Hockless, C. R.; Skelton, B. W.; Sobolev, A. N.; White, A. H. Tetrahedron 1994, 50, 941–956.
- 52. Boge, T. C.; Georg, G. I.; Tamariz, J. In Enantioselective Synthesis of β -Amino Acids. Juaristi, E., Ed.; Wiley-VCH: New York, 1997.
- 53. Hayashi, Y.; Katade, J.; Harada, T.; Tachiki, A.; Iijima, K.; Takiguchi, Y.; Muramatsu, M.; Miyazaki, H.; Asari, T.; Okazaki, T.; Dato, Y.; Yasuda, E.; Yano, M.; Uno, I.; Ojima, I. J. Med. Chem. 1998, 41, 2345–2360.
- 54. Martín-Vilà, M.; Muray, E.; Aguado, G. P.; Alvarez-Larena, A.; Branchadell, V.; Minguillón, C.; Giralt, E.; Ortuño, R. M. Tetrahedron: Asymmetry 2000, 11, 3569–3584.
- 55. Sabbioni, G.; Jones, J. B. J. Org. Chem. 1987, 52, 4565–4570.
- 56. Ibuka, T.; Akimoto, N.; Tanaka, M.; Nishii, S.; Yamamoto, Y. J. Org. Chem. 1989, 54, 4055–4061.
- 57. Muray, E.; Alvarez-Larena, A.; Piniella, J. F.; Branchadell, V.; Ortuño, R. M. J. Org. Chem. 2000, 65, 388-396.
- 58. Kennewell, P. D.; Matharu, S. S.; Taylor, J. B.; Westwood, R.; Sammes, P. G. J. Chem. Soc., Perkin Trans. 1 1982, 2563–2570.
- 59. Cannon, J. G.; Rege, A. B.; Gruen, T. L. J. Med. Chem. 1972, 15, 71–75.
- 60. Appella, D. H.; Chritianson, L. A.; Klein, D. A.; Richards, M. R.; Powell, D. R.; Gellman, S. H. J. Am. Chem. Soc. 1999, 121, 7574–7581.
- 61. Porter, E. A.; Wang, X.; Lee, H.-S.; Weisblum, B.; Gellman, S. H. Nature 2000, 404, 565.
- 62. Nöteberg, D.; Brånalt, J.; Kvarnström, I.; Classon, B.; Samuelsson, B.; Nillroch, U.; Danielson, H.; Karlén, A.; Hallberg, A. Tetrahedron 1997, 53, 7975–7984.
- 63. Nativ, E.; Rona, P. Isr. J. Chem. 1972, 10, 55.
- 64. LePlae, P. R.; Umezawa, N.; Lee, H.-S.; Gellman, S. H. J. Org. Chem. 2001, 66, 5629–5632.
- 65. Hünig, S.; Kahanek, H. Chem. Ber. 1953, 86, 518-522.
- 66. Moriconi, E. J.; Mazzochi, P. H. J. Org. Chem. 1966, 31, 1372–1379.
- 67. Appella, D. H.; Christianson, L. A.; Karle, I. L.; Powel, D. R.; Gellman, S. H. J. Am. Chem. Soc. 1999, 121, 6206-6212.
- 68. Prout, F. S.; Beaucaire, V. D.; Dyrkacz, G. R.; Koppes, W. M.; Kuznicki, R. E.; Marlewski, T. A.; Pienkowski, J. J.; Puda, J. M. J. Org. Chem. 1973, 38, 1512–1517.
- 69. Schultz, A. G.; Alva, C. W. Org. Synth. 1996, 73, 174–183.
- 70. (a) Barluenga, J.; Aznar, F.; Valdés, C.; Martín, A.; García-Granda, S.; Martín, E. J. Am. Chem. Soc. 1993, 115, 4403–4404. (b) Enders, D.; Meyer, O.; Raabe, G. Synthesis 1992, 1242–1244. (c) Barluenga, J.; Aznar, F.; Ribas, C.; Valdés, C. J. Org. Chem. 1997, 62, 6746–6753.
- 71. Carlson, P. H.; Katsuki, T.; Martín, V. S.; Sharpless, K. B. J. Org. Chem. 1981, 46, 3936–3938.
- 72. Kennewell, P. D.; Mathrau, S. S.; Taylor, J. B.; Westwood, R.; Sammes, P. G. J. Chem. Soc., Perkin Trans. 1 1982, 2553–2562.
- 73. Schroff, C. C.; Stewart, W. S.; Uhm, S. J.; Wheeler, J. W. J. Org. Chem. 1971, 36, 3356–3361.
- 74. Payne, G. B. J. Org. Chem. 1967, 32, 3351–3355.
- 75. Witiak, D. T.; Sinha, B. K.; Lee, O. S.; Feller, D. R. J. Med. Chem. 1972, 15, 803–808.
- 76. Kennewell, P. D.; Mathrau, S. S.; Taylor, J. B.; Westwood, R.; Sammes, P. G. J. Chem. Soc., Perkin Trans. 1 1982, 2563–2570.
- 77. Burgess, K.; Li, S.; Rebenspies, J. Tetrahedron Lett. 1997, 38, 1681–1684.
- 78. Atherton, E.; Sheppard, R. C. Solid Phase Peptide Synthesis, A Practical Approach. IRL: Oxford, 1989.
- 79. Goodman, M.; Shao, H. Pure Appl. Chem. 1996, 68, 1303–1308.
- 80. Davis, F. A.; Liu, H.; Reddy, G. V. Tetrahedron Lett. 1996, 37, 5473–5476.
- 81. Davis, F. A.; Liang, C.-H.; Liu, H. J. Org. Chem. 1997, 62, 3796–3797.
- 82. Fowden, L. Nature 1955, 176, 347–348.
- 83. Fowden, L. Biochem. J. 1956, 64, 323–332.
- 84. (a) Hamada, Y.; Shioiri, T. J. Org. Chem. 1986, 51, 5489–5490. (b) Noma, M.; Noguchi, M.; Tamaki, E. Tetrahedron Lett. 1971, 2017–2020.
- 85. Zagari, A.; Palmer, K. A.; Gibson, K. D.; Nemethy, G.; Scheraga, H. A. Biopolymers 1994, 34, 51–60.
- 86. Cromwell, N. H.; Phillips, B. Chem. Rev. 1979, 79, 331–358.
- 87. Hanessian, S.; Bernstein, N.; Yang, R.-Y.; Maguire, R. Bioorg. Med. Chem. Lett. 1999, 9, 1437–1442.
- 88. Seebach, D.; Vettiger, T.; Muller, H.-M.; Plattner, D. A.; Petter, W. Liebigs Ann. Chem. 1990, 687–695.
- 89. Anderson, Jr. A. G.; Lok, R. J. Org. Chem. 1972, 37, 3953–3955.
- 90. Gaertner, V. R. Tetrahedron Lett. 1966, 4691-4694.
- 91. (a) Barlow, D. J.; Thornton, J. M. J. Mol. Biol. 1988, 201, 601–619. (b) Koskinen, A. M. P.; Rapoport, H. J. Org. Chem. 1989, 54, 1859–1866. (c) Ibrahim, H. H.; Lubell, W. D. J. Org. Chem. 1993, 58, 6438-6441.
- 92. Bashyal, B. P.; Fleet, G. W. J.; Gough, M. J.; Smith, P. W. Tetrahedron 1987, 43, 3083–3093.
- 93. Moss, W. O.; Bradbury, R. H.; Hales, N. J.; Gallagher, T. J. Chem. Soc., Chem. Commun. 1990, 51–53.
- 94. Seebach, D.; Grobel, B.-Th.; Beck, A. K.; Braun, M.; Geiss, K.-H. Angew. Chem. Int. Ed. Engl. 1972, 11, 443–444.
- 95. Moss, W. O.; Wakefield, E.; Mahon, M. F.; Molloy, K. C.; Bradbury, R. H.; Hales, N. J.; Gallagher, T. Tetrahedron 1992, 48, 7551–7564.
- 96. Corey, E. J.; Beames, D. J. J. Am. Chem. Soc. 1973, 95, 5829–5831.
- 97. Stork, G.; Zhao, K. Tetrahedron Lett. 1989, 30, 287-290.
- 98. Udding, J. H.; Tuijp, C. J. M.; Hiemstra, H.; Speckamp, W. N. Tetrahedron 1994, 50, 1907–1918.
- 99. Beck, A. L. J. Tetrahedron 1981, 37, 3073-3100.
- 100. Esch, P. M.; Hiemstra, H.; de Boer, R. F.; Speckamp, N. Tetrahedron 1992, 48, 4659–4676.
- 101. Esch, P. M.; Boska, I. M.; Hiemstra, H.; de Boer, R. F.; Speckamp, N. Tetrahedron 1991, 47, 4039–4062.
- 102. Souers, A. J.; Ellman, J. A. J. Org. Chem. 2000, 65, 1222–1224.
- 103. Varray, S.; Gauzy, C.; Lamaty, F.; Lazaro, R.; Martinez, J. J. Org. Chem. 2000, 65, 6787–6790.
- 104. Ohtani, B.; Tsuru, S.; Nishimoto, S.; Kagiya, T. J. Org. Chem. 1990, 55, 5551–5553.
- 105. Horikawa, M.; Busch-Petersen, J.; Corey, E. J. Tetrahedron Lett. 1999, 40, 3843–3846.
- 106. Hu, L.-Y.; Ryder, T. R.; Nikam, S. S.; Millerman, E.; Szoke, B. G.; Rafferty, M. F. Bioorg. Med. Chem. Lett. 1999, 9, 1121–1126.
- 107. (a) Giannis, A.; Kolter, T. Angew. Chem., Int. Ed. Engl. 1993, 32, 1244–1267. (b) Burger, K.; Mutze, K.; Hollweck, W.; Koksch, B. Tetrahedron 1998, 54, 5915–5928.
- 108. Hao, J.; Ohkura, H.; Amii, H.; Uneyama, K. Chem. Commun. 2000, 19, 1883–1884.
- 109. Osipov, S. N.; Bruneau, C.; Picquet, M.; Kolomiets, A. F.; Dixneuf, P. H. Chem. Commun. 1998, 2053–2054.
- 110. Seebach, D.; Dziadulewicz, E.; Bhrendt, L.; Cantoreggi, S.; Fitzi, R. Liebigs Ann. Chem. 1989, 1215–1232.
- 111. Mehlführer, M.; Thirring, K.; Berner, H. J. Org. Chem. 1997, 62, 4078–4081.
- 112. Kende, A. S. Organic Reactions, Adams, R., Blatt, A. H.,

Boekelheide, V., Cairns, T. L., Curtin, D. Y., Niemann, C., Eds.; Wiley: New York, 1960; Vol. 11, pp 261–316.

- 113. (a) Nagasawa, H. T.; Elberling, J. A. Tetrahedron Lett. 1966, 5393–5399. (b) Nagasawa, H. T.; Elberling, J. A.; Fraser, P. S. J. Med. Chem. 1971, 14, 501–505.
- 114. Creighton, C. J.; Reitz, A. B. Org. Lett. 2001, 3, 893–895.
- 115. Fink, B. E.; Kym, P. R.; Katzenellenbogen, J. A. J. Am. Chem. Soc. 1998, 120, 4334–4344.
- 116. Nadin, A.; Derrer, S.; McGeary, R. P.; Goodman, J. M.; Raithby, P. R.; Holmes, A. B.; O'Hanlon, P. J.; Pearson, N. D. J. Am. Chem. Soc. 1995, 117, 9768–9769.
- 117. (a) Pfaff, M.; Tangemann, K.; Müller, B.; Gurrath, M.; Müller, G.; Kessler, H.; Timpl, R.; Engel, J. J. Biol. Chem. 1994, 269, 20233–20238. (b) Geyer, A.; Müller, G.; Kessler, H. J. Am. Chem. Soc. 1994, 116, 7735–7743.
- 118. Cabrele, C.; Langer, M.; Beck-Sickinger, A. G. J. Org. Chem. 1999, 64, 4353–4361.
- 119. Beck-Sickinger, A. G.; Jung, G. Biopolymers 1995, 37, 123–142.

Biographical sketch

Kyung-Ho Park was born in 1960 in Kyungpook Cheongdo (Korea). He studied chemistry at Kyungpook National University in Korea (1988), and worked at Korea Research Institute of Chemical Technology (KRICT) until 1995. He obtained his PhD degree on combinatorial synthesis of hydantoins with isoxazoline heterocycle in Professor Mark J. Kurth's group at the University of California, Davis (2000). He joined DuPont CR&D as a research chemist in 2000. His current research interests focus on lead discovery of biologically active small organic molecules as well as new materials via solid-phase combinatorial chemistry.

Mark J. Kurth was born in Iowa in 1953. He received his B.A. degree from the University of Northern Iowa in 1976 where he worked with Professor James G. MacMillan. His PhD degree was from the University of Minnesota (1980) where he worked with Professor Thomas R. Hoye on the total synthesis of brominated marine natural products such aplysistatin. After a postdoctoral study with Professor Wolfgang Oppolzer at the University of Geneva (1980–1981) developing chiral auxiliaries for the Diels–Alder reaction, he joined the chemistry faculty at the University of California/Davis. His current research activities span natural products total synthesis, synthetic methods development, solid-phase organic synthesis and combinatorial chemistry, and polymer synthesis and characterization.